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***The Effects of Purity and Pressure on
Hydrogen Embrittlement of Metallic
Materials (ID 149)***

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- **Introduction to Hydrogen Embrittlement (HE)**
 - ✓ **Causes and Mechanisms**
 - ✓ **Scope of Research**
- **Results and Data**
 - ✓ **Test Methods**
 - ✓ **Effects of Pressure**
 - ✓ **Effects of H₂ gas purity**
- **Issues to Address**

Hydrogen Embrittlement in Steels

- **Loss of ductility and deformation capacity in the presence of hydrogen**
- **Strongly affects high-strength steels**
- **Maximum embrittlement at room temperature (~20°C)**
- **Causes hydrogen transport by dislocations**

Hydrogen Embrittlement in Steels

■ Factors affecting HE

- ✓ Environment
- ✓ Material properties and surface condition

■ Possible Mechanisms of HE

- ✓ Stress-induced hydride formation
- ✓ Hydrogen-enhanced localized plasticity (HELP)
- ✓ Hydrogen-induced decohesion

- **Gas purity**
- **Pressure**
- **Temperature**
- **Exposure time (affects diffusion in internal HE)**
- **Stress and strain rates**

Material Properties related to HE

- **Chemical composition of metal**
- **Heat treatment, welding**
- **Microstructure**
- **Cracks, corrosion pits, and other surface defects**

■ Metals

- ✓ **Carbon, Low alloy, and Stainless Steels**
- ✓ **Aluminum and Copper**

■ Focusing on three aspects of HE for steels

- ✓ **Effective testing methods for HE**
- ✓ **Effects of H₂ gas pressure (700-1000 bar)**
- ✓ **Effects of H₂ gas purity**

Hydrogen Embrittlement

- **Austenitic alloys suffer less embrittlement than ferritic alloys**
- **Martensitic specimens are very sensitive to HE**
- **Steels often become less ductile, but strength is not significantly reduced by HE**
- **Aluminum and copper alloys have shown high resistance to HE in tensile testing**

TESTING METHODS

Testing for HE

- **Need to simulate in-service stresses of pressure vessels -external HE effects**
- **High sensitivity**
- **Capable of being reproduced**
- **Small cells for lower cost and easy cleaning**

- **Compare behavior under pressurized hydrogen vs. inert gas**
- **Provide data for changes in ductility-elongation and %RA**

$$\% RA = \frac{A_i - A_f}{A_i} \times 100\%$$

- **Tensile stresses are uniaxial**

■ Wedge Opening Load (WOL)

- ✓ Test of threshold stress intensity factor, K_{TH}
- ✓ Crack growth Maximum acceptable crack growth: 0.25mm [1]

■ Compact Tension (CT)

- ✓ Fatigue crack growth rate, da/dN
- ✓ K_{TH} Acceptable criteria: $K_{TH} > (60/950) \times R_m$ (MPa-m^{0.5}), where R_m is UTS of metal [1]
- ✓ Plane-strain fracture toughness, K_{IC}

- **Disk rupture**
 - ✓ **Provides strength comparison in H₂ and He environments**
 - ✓ **Creates triaxial stress state**

- **Delayed rupture**

- **Disk fatigue test**
 - ✓ **Good for simulating life of a pressure vessel**

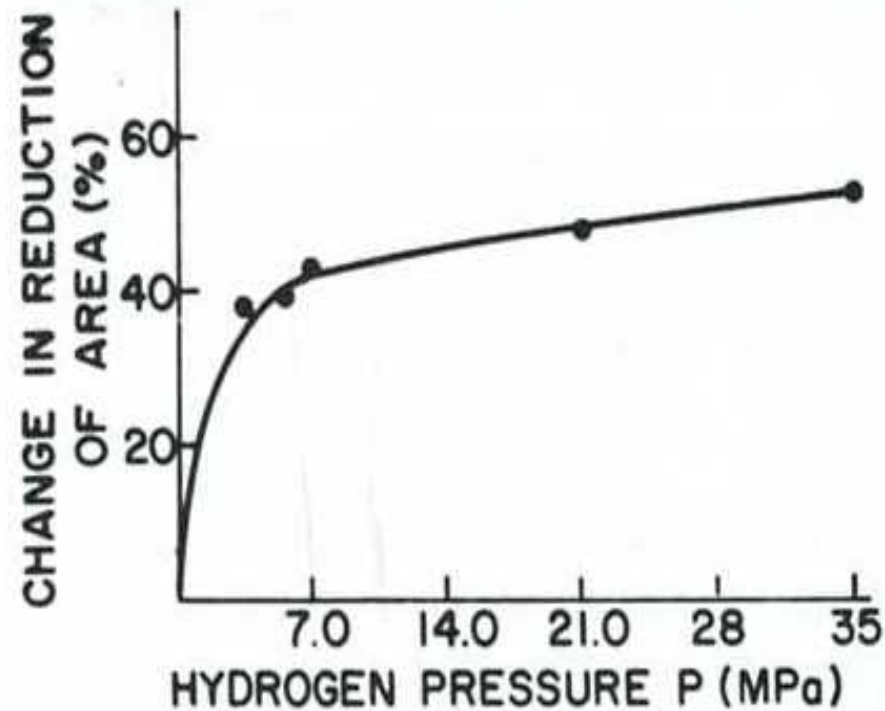
EFFECTS OF HYDROGEN GAS PRESSURE

- **HE generally increases with partial hydrogen pressure**

- **Some tests showed maximum HE at a certain pressure level**
 - ✓ **~100 bar for carbon and low alloys where UTS < 1000 MPa**
 - ✓ **~25 bar for AISI 321 stainless steel [2]**

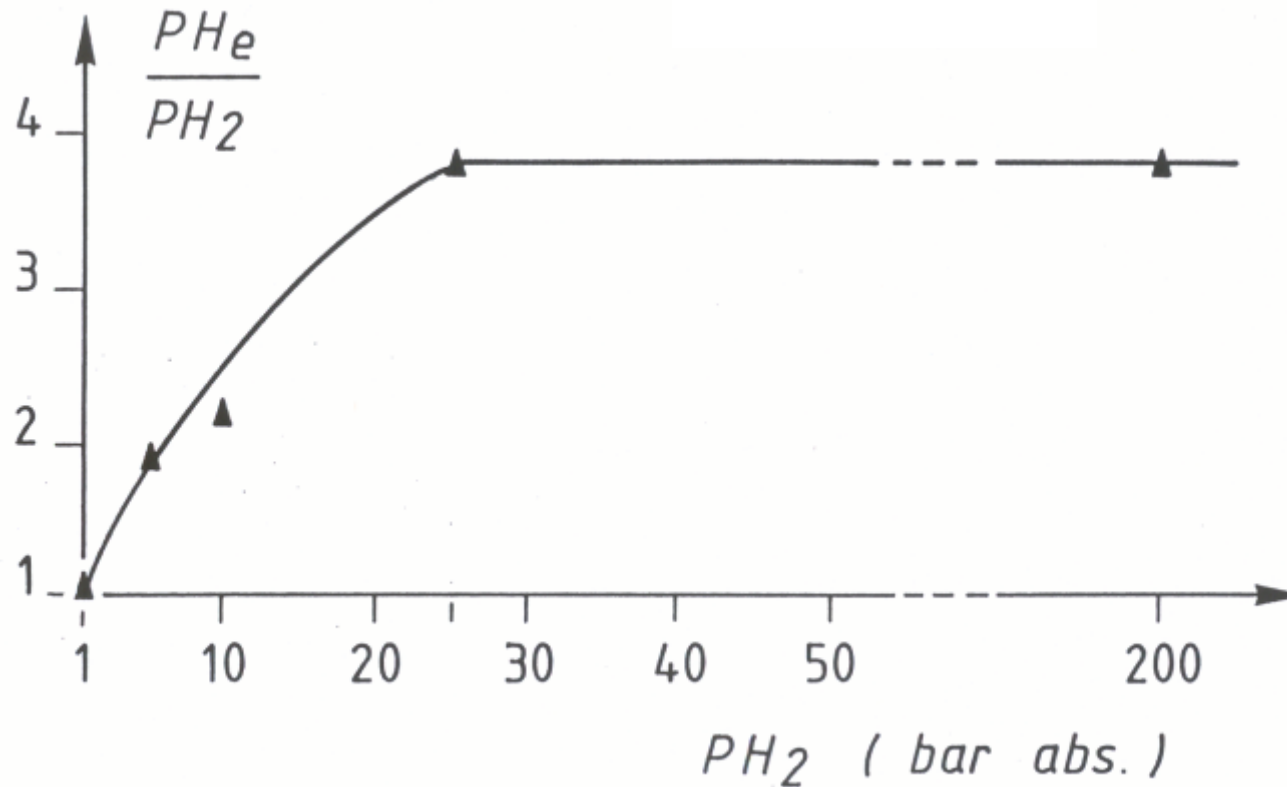
HE Test Results

HEE observed in double-notched tensile specimens.



Large ductility loss with increase of hydrogen pressure for carbon steel [3]

- **Hydrogen pressure**



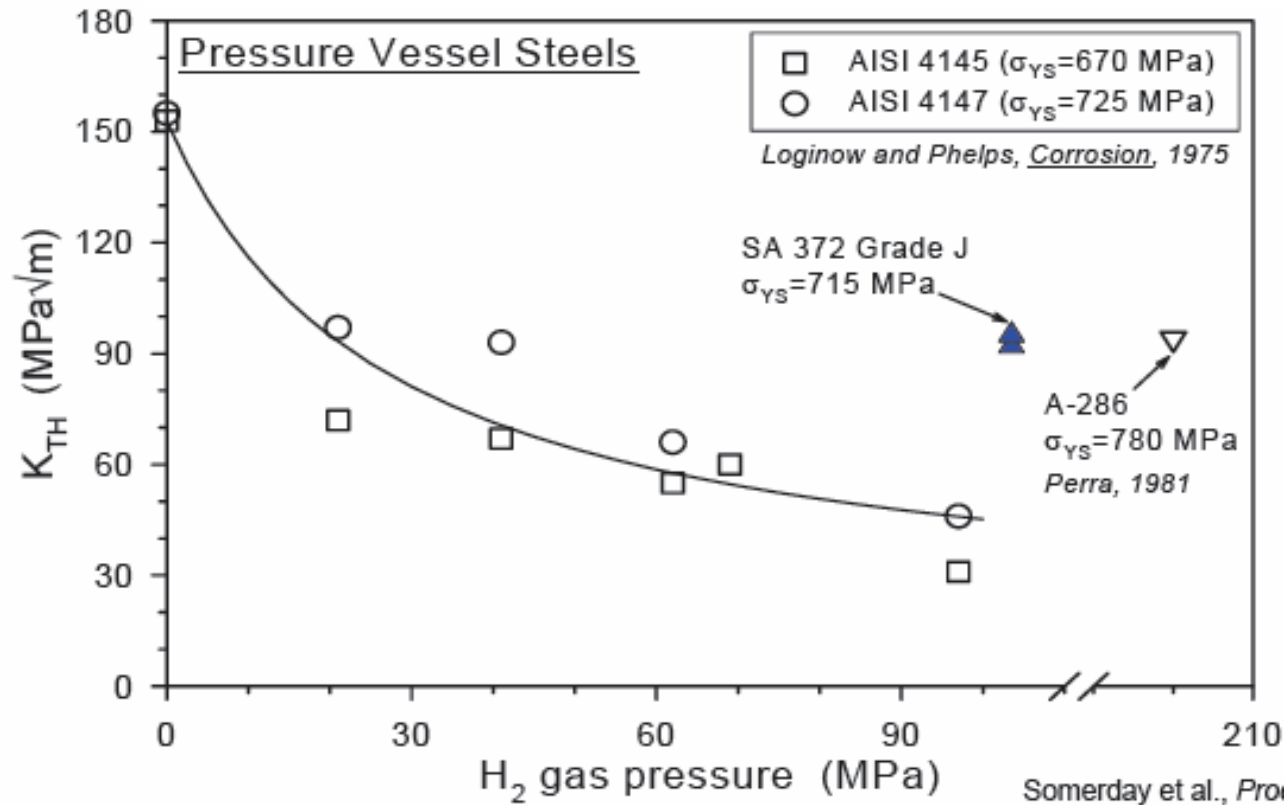
**Influence of H_2S partial pressure
for AISI 321 steel**

316 Stainless Steel

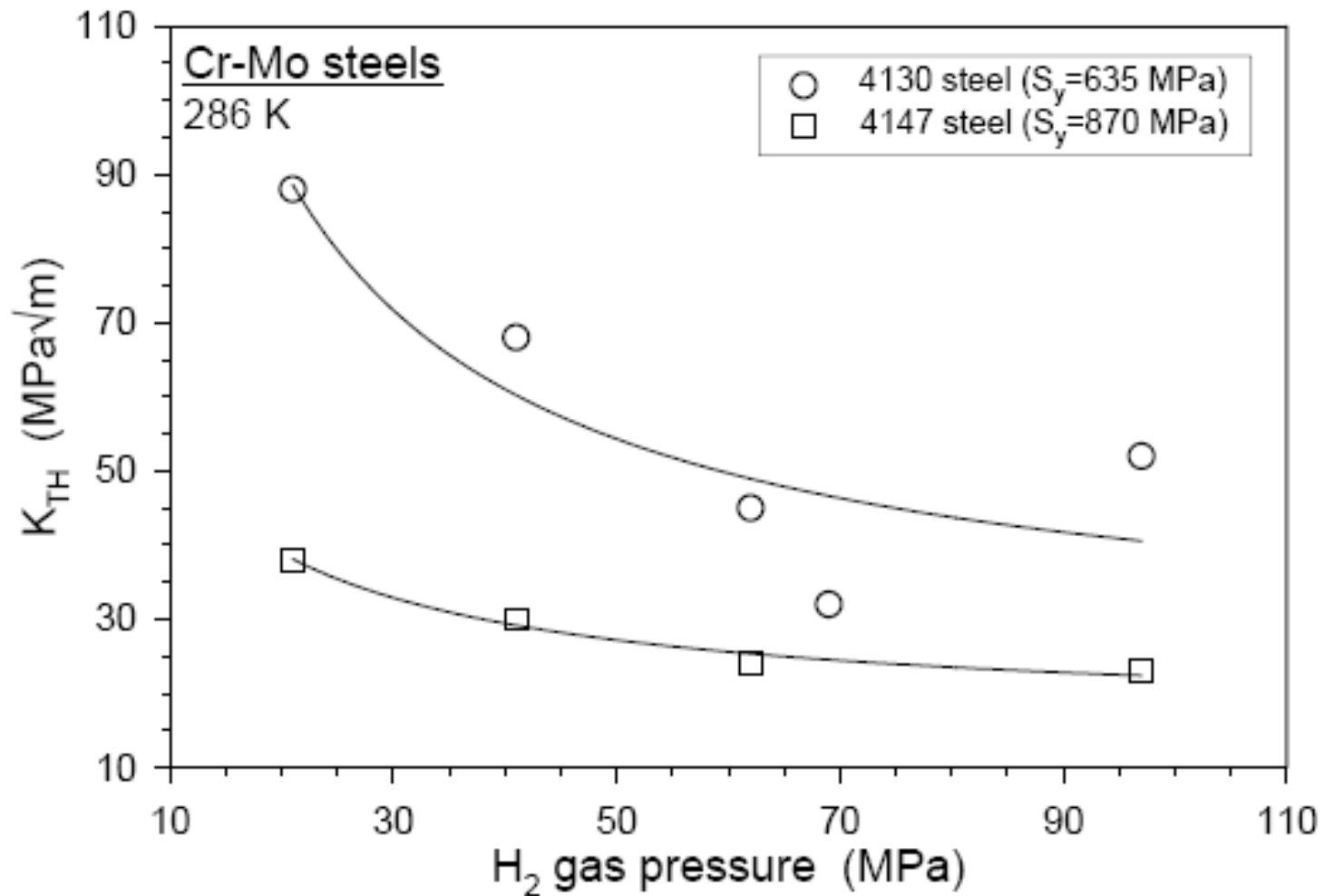
Material	Thermal precharging	Test environment	Strain rate (s ⁻¹)	S _y (MPa)	S _u (MPa)	El _u (%)	El _t (%)	RA (%)
Not specified	None	69 MPa He	---	214	496	---	68	78
	None	69 MPa H ₂		214	524	---	72	77
Cold drawn rod, heat W69	None	69 MPa He	0.67 x10 ⁻³	441	648	---	59	72
	None	69 MPa H ₂		---	683	---	56	75
Annealed plate, heat O76	None	Air	3 x10 ⁻³	262	579	---	68	78
	None	69 MPa H ₂		221	524	---	72	77
Annealed sheet	None	Air	0.6 x10 ⁻³	263	568	---	90	75
	None	70 MPa He		248	565	---	85	70
	None	70 MPa H ₂		249	566	---	85	75

Successful performance of 316 steel in tensile tests with high pressure hydrogen [4]

Pressure Test Results

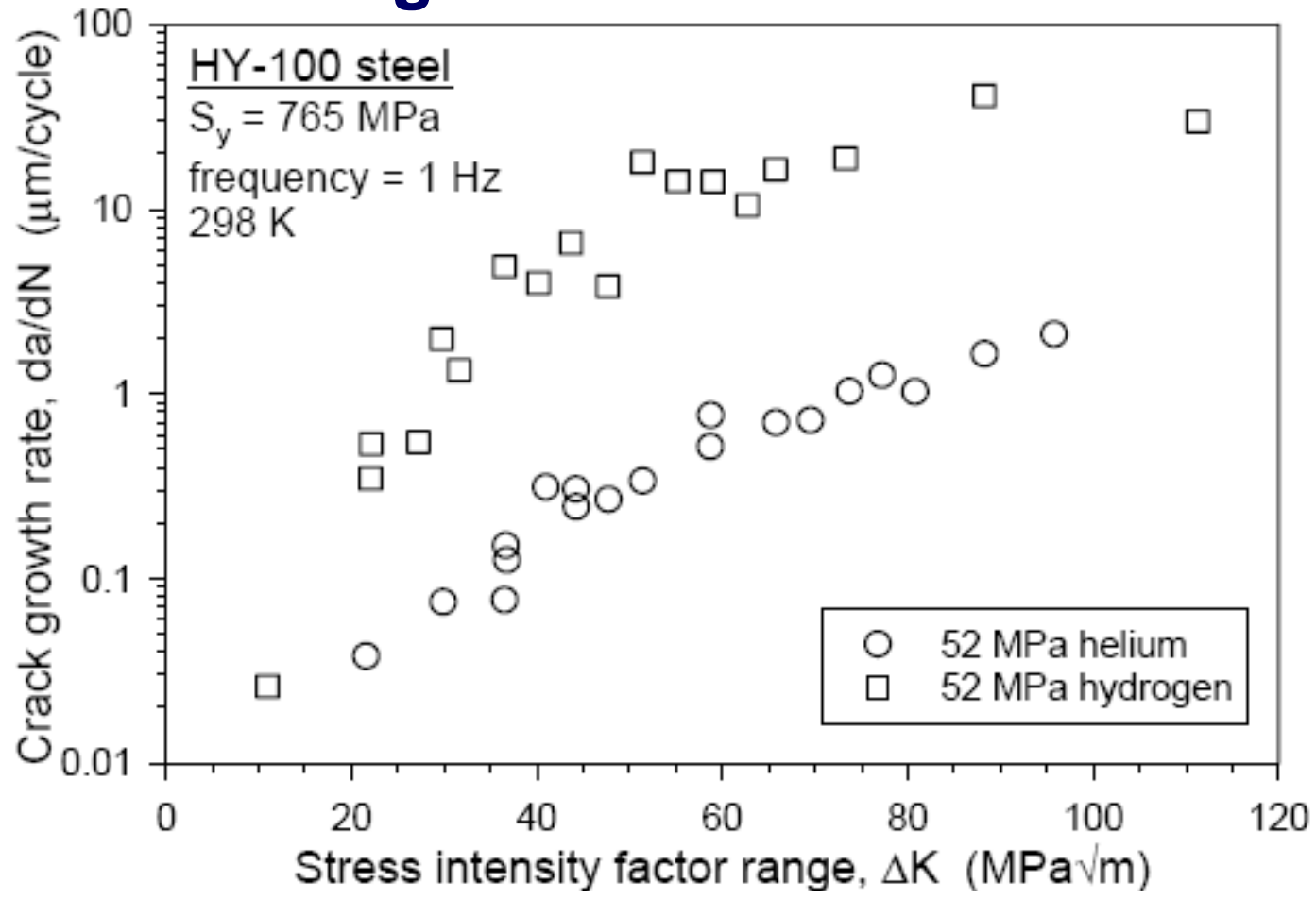


Cracking threshold significantly decreases as hydrogen pressure increases for low alloy steel [4]

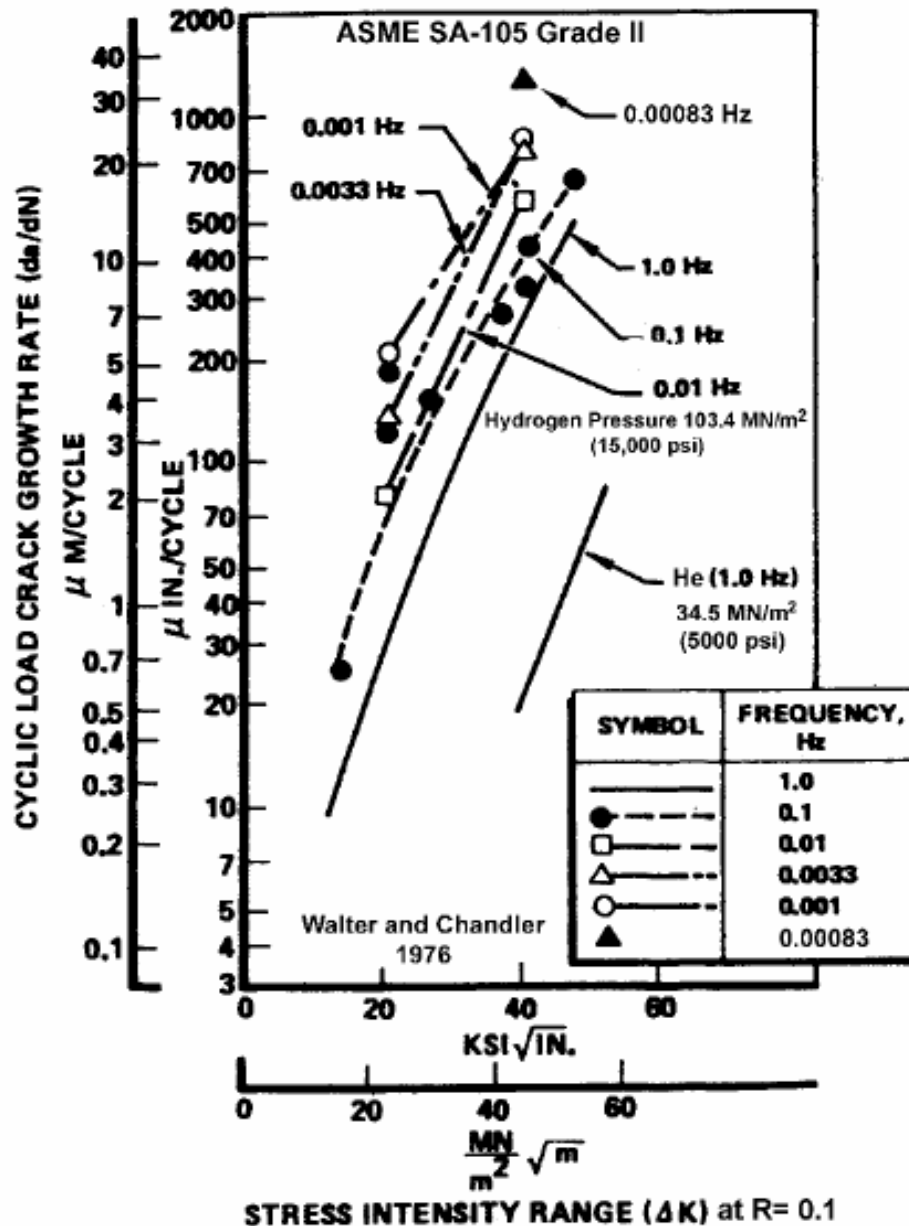


HE effects on type 4147 appear to level off at pressures higher than 60 MPa [4]

Effects of 52 MPa hydrogen on fatigue crack growth for HY-100 Steel



HY-100 showed significantly increased crack growth rate in 52 MPa hydrogen [4]



In tests performed at 103.4 MPa on SA-105 steel, fatigue crack growth was slower at higher frequencies [5]

Results of High Pressure Tests

- **Losses in ductility increase with pressure, although several steels reached maximum embrittlement at a threshold pressure**
- **Fracture toughness and resistance to crack propagation decrease**
- **Strength of the material is usually not significantly affected**

Effects of High Pressure Hydrogen

- **Fatigue resistance decreases at higher pressures**
- **A286 and 316 stainless steels have shown the most resistance to HE**
- **Aluminum and copper alloys appear to be resistant to HE**
- **More fatigue testing at high pressure needs to be performed on these materials**

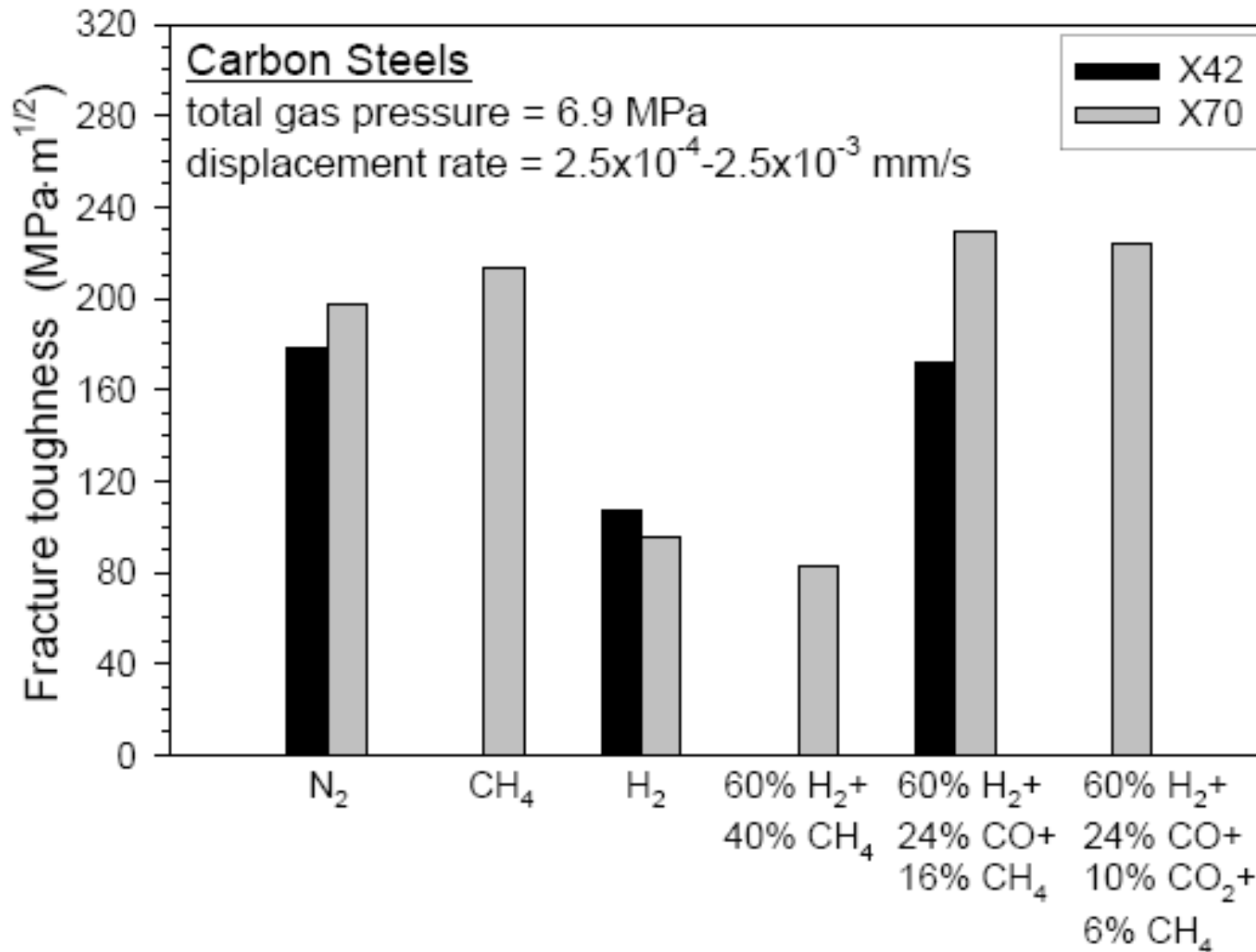
EFFECTS OF HYDROGEN GAS PURITY

Impurities Affecting HE

HE Inhibitor	No Effect	Embrittling Effect
O_2	CH_4	H_2S
SO_2	N_2	CO_2

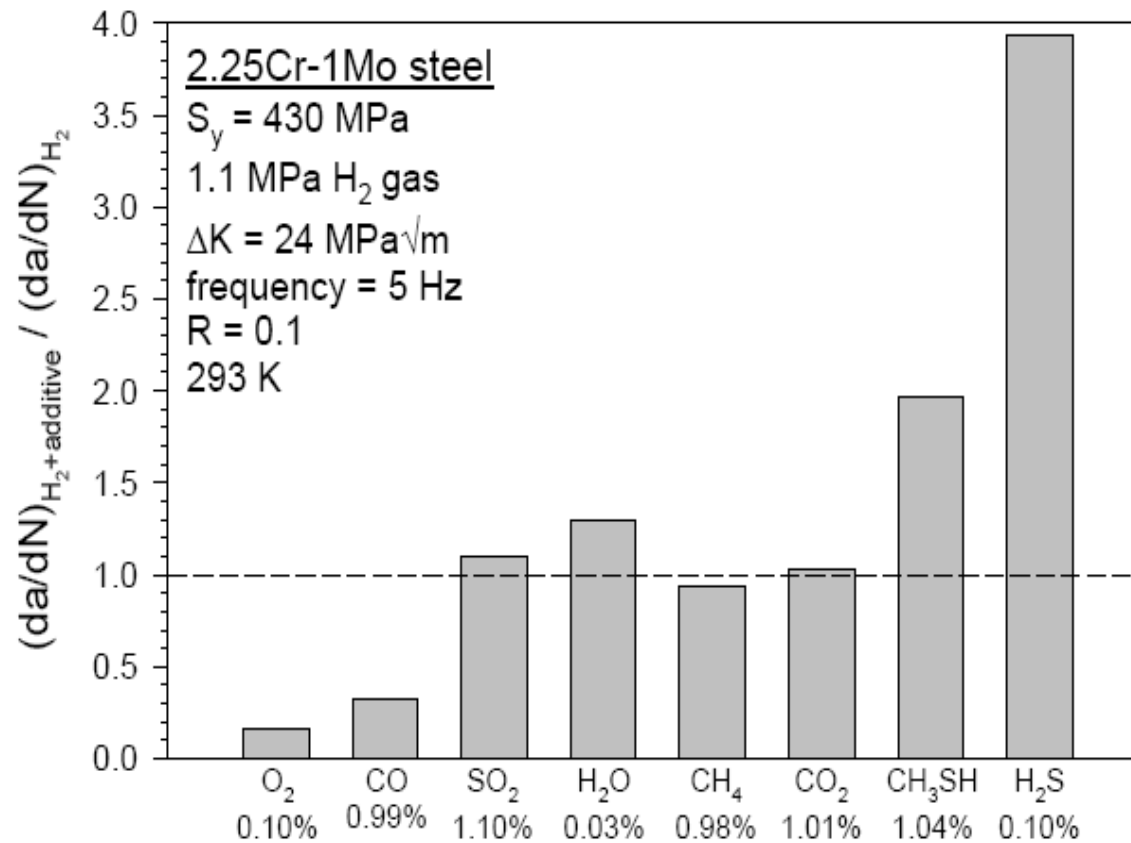
H_2O has demonstrated both embrittling and inhibiting effects

- **Oxygen has shown inhibiting effects in delayed disk rupture**
- **Varying results for impurities such as CH₄ and CO₂**
- **H₂S has consistently accelerated HE**



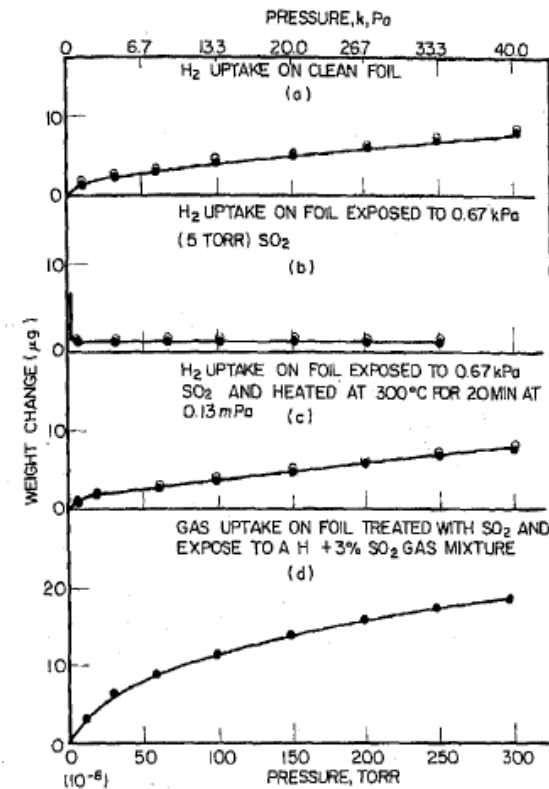
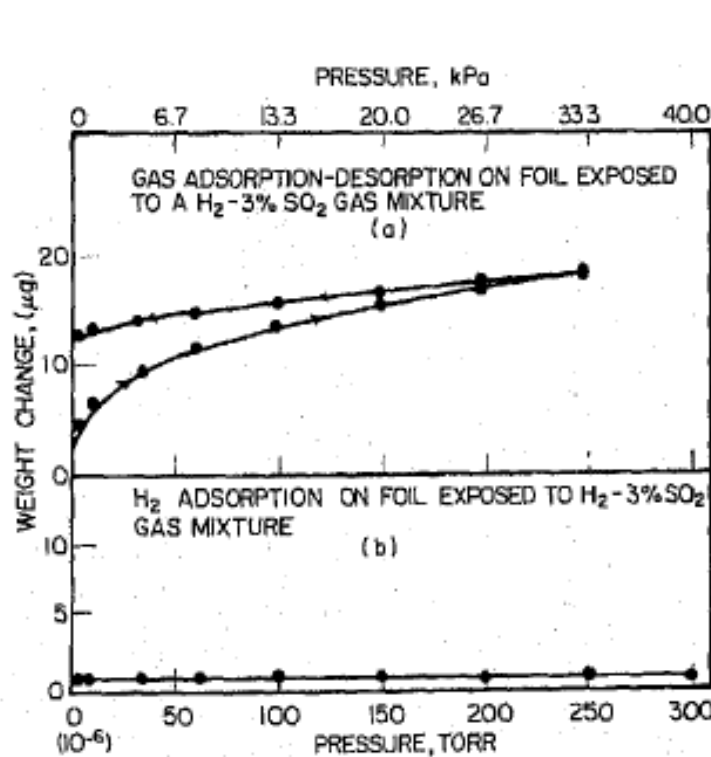
Effects of gas impurity on HE for carbon steels [4]

Impurity effects on fatigue crack growth



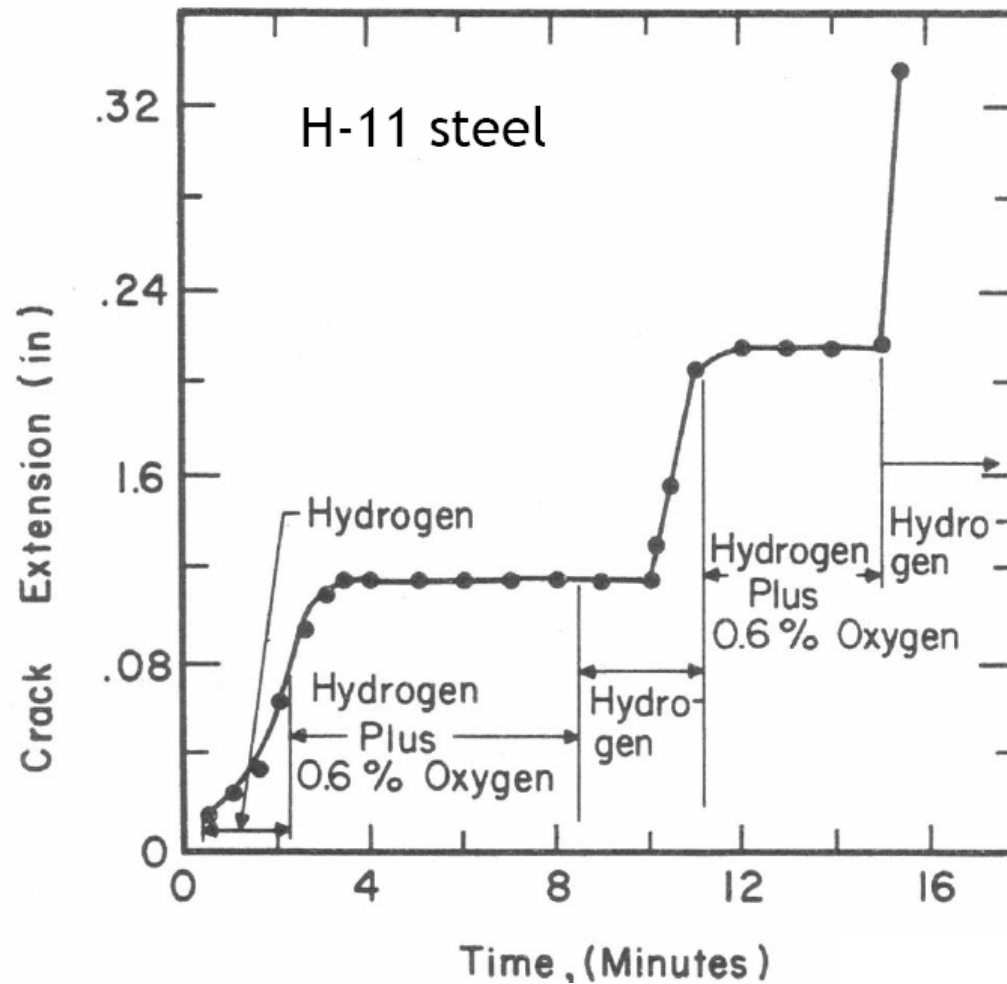
Comparison between pure gas and H_2 with additives [4]

Inhibiting effects of SO₂



Sulfur dioxide exhibited inhibitory effects as a pretreatment for steel sheet [6]

Inhibiting effects of oxygen on crack growth in H-11 steel



Results from Impurity Tests

- **O₂ consistently acts as an inhibitor**
- **H₂S has consistently accelerated HE**
- **Pretreating with SO₂ had inhibiting effects during pretreatments**
- **More data and information is needed regarding purity at higher pressures**

Areas for Further Research

- **More information needed about threshold pressures at which maximum embrittlement occurs**
- **Fatigue data for metals demonstrating good HE resistance (A286, 316, Al and Cu alloys)**
- **More information needed about effects of inhibitors**
 - ✓ **Resolve conflicting claims**
 - ✓ **Specific concentrations for mixtures**
 - ✓ **Inhibiting at higher pressures**

References

1. ISO 11114-4: Transportable gas cylinders, compatibility of cylinder and valve materials with gas contents, 2006.
2. Barthelemy, H. Compatibility of Metallic Materials with Hydrogen. Air Liquide.
3. Gutierrez-Solana, F. and M. Elices. High Pressure Hydrogen Behavior of a Pipeline Steel, ed. by C. G. Interrante and G. M. Pressouyre. Proceedings of the First International Conference on Current Solutions to Hydrogen Problems in Steels, November 1 (1982), pp. 181-185.
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5. Lam, P S., R. L. Sindelar, and T. M. Adams. Literature Survey of Gaseous Hydrogen Effects on the Mechanical Properties of Carbon and Low Alloy Steels. Savannah River National Laboratory, ASME Pressure Vessels and Piping Division Conference, 22 July 2007.
6. Srikrishnan, V. and P. J. Ficalora. The Role of Gaseous Impurities in Hydrogen Embrittlement of Steel. September 7, 1976.
7. Balch, D., San Marchi, C., and B. Somerday. Hydrogen-Assisted Fracture: Materials Testing and Variables Governing Fracture. Sandia National Laboratories. Hydrogen Pipeline Working Group Workshop, August 30-31, 2005.